

GGLOVENKO, Nikolay Aleksandrovich; YAROTSKIY, Aleksey Samoylovich;  
DMITRENKO, N.Z., red.; POLONSKIY, S.A., tekhn. red.

[Planning capital investments and unfinished construction in the  
industry of the Moldavian S.S.R.] Planirovaniye kapital'nykh vlo-  
zhenii i nezavershennoe stroitel'stvo v promyshlennosti Moldav-  
skoi SSR (po dannym 16 stroek). Kishinev, Izd-vo "Shtiintsa,"  
1962. 32 p. (MIRA 15:12)

(Moldavia—Capital investments)  
(Moldavia—Construction industry—Management)

BORTNIKOV, V.B., kand. ekon. nauk, red.; MEDNEK, V.P., red.; FEDOTOVA, R.D., red.; DMITRENKO, N.Z., red.; POLONSKIY, S.A., tekhn.red.

[Problems of the economics of capital construction in the Moldavian S.S.R.] Voprosy ekonomiki kapital'nogo stroitel'stva v Moldavskoi SSR; materialy. Kishinev, Shtiintsa, 1962. 145 p.  
(MIRA 16:2)

1. Nauchno-ekonomiceskaya konferentsiya po stroitel'stvu v Moldavskoy SSR, Kishinev, 1961. 2. Zamestitel' predsedatel' Gosudarstvennogo komiteta Soveta Ministrov SSSR po delam stroitel'stva Moldavskoy SSR (for Mednek). 3. Zaveduyushchiy sektorom ekonomiki stroitel'noy industrii Instituta ekonomiki Akademii nauk Moldavskoy SSR(for Bortnikov).

(Moldavia--Construction industry--Management)

KLOCHKO, Grigoriy Sidorovich; DMITRENKO, N.Z., red.; MARKOVICH,  
G.L., tekhn. red.

[How to increase the profitability of the communal economy of  
collective farms] Puti povysheniia tovarnosti obshchestvennogo  
khoziaistva kolkhozov. Kishinev, Izd-vo "Shtiintsa" Akad. nauk  
Moldavskoi SSR, 1962. 226 p.  
(MIRA 15:6)  
(Collective farms)

CHIKRYZOVA, Yo.G., red.; INALIKOV, Yu.S., red.; LIPIS, B.V., red.;  
DMITREJKO, N.Z., red.; SHCHETININA, Ye.A., red.; LEDTICE,  
E.M., tekhn. red.

[Theory and practice of polarographic analysis] Teoriia i praktika poliarograficheskogo analiza; materialy. Kishinev, Izd-vo "Shtiintsa" Akad. nauk Moldavskoi SSR, 1962. 425 p.

(MIRA 15:12)

1. Vsesoyuznoye soveshchaniye po poliarograficheskому analizu.  
1st, 1959.

(Polarography--Congresses)

KARZIN, V. A.; VASIL'YEV, P. S.; DMITRENKO, O. I.

Moscow

Laboratory of Colloidal Chemistry, Physico-Chemical Institute imeni L. Ya. Karpov,  
(-1940-).

"The Effect of the Solubility of Silver Salts on their Adsorption by Mixed Gels of  
Silicic Acid and Sesquioxides."

Zhur. Fiz. Khim. Vol. 14, No. 12, 1940.

Dmitrenko, O. I.

RA 2/50T38

USER/Chemistry - Colloids  
Literature

Sep/Oct 48

"Answer to Ye. N. Gapon's Article," O. I.  
Dmitrenko, 1 p

"Kolloid Zhur" Vol X, No 5

Written in answer to a criticism by Gapon of a previous article by Dmitrenko and M. M. Kolesnikov, "Adsorption Properties of Highly Refined Organic Mineral Gels," in "Kolloidnyy Zhurnal," No. 8, 1945, p 319. Dmitrenko says Gapon attacked individual points, but not principal thesis of initial article, which stated that

2/50T38

USSR/Chemistry - Colloids  
Literature (Contd)  
Sep/Oct 48

"Introduction of organic substances (sodium humate) does not increase the acid or exchange properties of highly refined mineral gels." Submitted 8 Apr 48.

2/50T38

CH

2

*Molecular uptake of silver salts by iron hydroxide.* O. I. Dunitrenko, V. A. Kargin, and A. A. Ryabikin. *Vadodar. ZAMP*, 19, 8 19(1961). Gels obtained by hydrolyzing  $\text{FeCl}_3$  in  $\text{H}_2\text{O}$  were dialyzed and electrodialyzed (the nature of the membrane is not stated) for months. The elec. resistance  $R$  of the gels increased during this treatment, e.g., from 8 (gel A) to 188 (gel B) and then remained almost const. (C);  $R$  at distl.  $\text{H}_2\text{O}$  was 70 in these units. These gels were titrated potentiometrically with 0.01 N  $\text{AgNO}_3$  and a Ag electrode, and the no.  $t'$  of g.-equiv. of Ag taken up by 100 g. dry  $\text{Fe(OH)}_3$  was calc'd.  $t'$  was, e.g., 13, 82, and 12 for A, B, and C, resp., when the equil. concn. of  $\text{AgNO}_3$  was 0.01 N. The uptake by A was due to  $\text{Cl}^-$  in it, the uptake by B to  $\text{OH}^-$  (which substitutes for  $\text{Cl}^-$  during dialysis) and to true mol. adsorption, and the uptake by C was small because the sample was partly cryst. and the electrolytes were removed from the surface of the micelles. Gels prepd. by hydrolysis of  $\text{Fe}(\text{NO}_3)_3$  behaved similarly, and the uptake of  $\text{Ag}^+$  by fully dialyzed gels was equal to that by C; this showed that all  $\text{Cl}^-$  and  $\text{NO}_3^-$  were washed out. The uptake of Ag by aged gels from either  $\text{FeCl}_3$  or  $\text{Fe}(\text{NO}_3)_3$  decreased during dialysis (to stage B) but increased again to stage C. At stage C,  $t'$  was about 18 for 0.012 N  $\text{AgNO}_3$ .

J. J. Bikerman

1951

CA

Artificial weathering and synthesis of minerals during  
electrodialysis. O. I. Dymirekha and V. A. Kargin.  
*Khimi. Zhur.* 15, 289-304 (1951).—A method is given for  
studying reactions between 2 very dil. solns. without using  
large vols. A 5-chamber electrodialyzer is used. It is  
made up of anodic compartment|membrane|powl. mineral  
A in H<sub>2</sub>O|membrane|H<sub>2</sub>O|membrane|powl. mineral  
B in H<sub>2</sub>O|membrane|cathodic compartment. Cations  
from A and anions from B meet in the central compartment  
and thus react in a small vol. of soln. When both A and B  
were beidellite ("gumbris" from Ural), a mineral similar to  
steatite formed after 8 months in the central chamber;  
its  $\mu$  was 1.488. The anodic compartment contained  
calcite and unidentified minerals. When A and B were  
phlogopite, the central chamber after 10 days contained ap-  
parently magnesiohornblende, a 1.622; beidellite or potas-  
siophyllite, a 1.637; and saussite, a 1.811. The pH of the  
cathodic liquid was 7.6 and of the anodic 3.6. This method  
presumably reproduces, in a shorter time, the hydrolysis of  
minerals in the earth's crust. Electrodialysis of beidellite  
in the conventional 3-chamber app. yielded calcite. Elec-  
trodialysis of Fe(OH)<sub>3</sub> sol for 60 weeks gave goethite. J. J. Bikerman

Div. of Sedimentary Petrography, Inst. Geology AS USSR

~~DMITREY~~

Effect of accompanying electrolytes on the competitive adsorption of silver salts by sulfide gels of silicic acid and oxides of iron and aluminum. O. I. Dmitrenko and A. A. Ryabina (Geol. Inst., Acad. Sci. Ukr. SSR). *Kolloid. Ztschr.* 15, 29-34 (1953); cf. C.A. 45, 3083a. More  $\text{Ag}^+$  was adsorbed from an acetate buffer (pH 4.03) than from a  $\text{AgNO}_3$  soln. at the same pH. This was due to formation of  $\text{AgOAc}$  which is less sol. and hence better adsorbed than  $\text{AgNO}_3$ . The adsorption from a soln. of  $\text{AgOAc}$  only was even greater because in the reaction between  $\text{AgNO}_3$  and acetate buffer  $\text{HNO}_3$  formed. Adsorption of  $\text{Ag}_2\text{SO}_4$  is greater still. This inverse relation between adsorption and solv. shows that mols. rather than ions are adsorbed.  $\text{Fe}_2\text{O}_3 \cdot 2\text{SiO}_2$  and  $\text{Al}_2\text{O}_3 \cdot 2\text{SiO}_2$  were used as adsorbents. The latter compd. was prep'd. from  $\text{Al}(\text{NO}_3)_3$ ,  $\text{AlCl}_3$ , or  $\text{Al}_2(\text{SO}_4)_3$ , and the adsorptive properties of the 2 specimens were identical. Mol. adsorption is possible also on impure surfaces, but the extent of adsorption is affected by electrodialysis. J. J. Bikerman

11-11-54 ref

DMITRENKO, O.I.; ZHUPAKHINA, Ye.S.

Measuring soil pH under normal moisture conditions by using an  
unbreakable glass electrode [with summary in English]. Pochvovedenie  
(MLRA 10:5)  
no.1:111-123 Ja '57.

1. Institut okeanologii AN SSSR.  
(Soil acidity)

69-58-2 -6/23

AUTHOR:

Dmitrenko, O.I.

TITLE:

The Binding of Water by Finely Dispersed Precipitates  
1. Calculation of the Amount of Bound Water From an Electrolyte Concentration in a Solution at Equilibrium (Svyazyvaniye vody tonkodispersnymi osadkami 1. Vychisleniye kolichestva svyazannoy vody po kontsentratsii elektrolitov v ravnovesnom rasvore)

PERIODICAL:

Kolloidnyy zhurnal, 1958, Vol XX, Nr 2, pp 163-169 (USSR)

ABSTRACT:

During the interaction of finely dispersed precipitates with solutions of electrolytes, a simultaneous adsorption of the molecules of the solutes and the solvent takes place. The adsorption of the solutes is more completely investigated than the adsorption of the solvent. In this article, a method for the fast determination of the bound water in natural precipitates is given. Finely disperse siltclay and clay precipitates of the Bering Sea are used in the experiments. These precipitates are characterized by a high (up to 90%) content of pelitic fraction with a particle diameter of less than 0.01 mm. The amount of bound water was determined by: 1) the general humidity of the precipitates dried

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69-58-2 -6/23

The Binding of Water by Finely Dispersed Precipitates 1. Calculation of  
the Amount of Bound Water From an Electrolyte Concentration in a Solution  
at Equilibrium

in the thermostat at a temperature of 110°C; 2) the quality  
of washed out chlorine found during the washing of the pre-  
cipitates by the electrolyte solution, which contains no chlo-  
rine ions; 3) the chlorine concentration in the soil solu-  
tion. The layers of water adjacent to the surface of the  
colloidal particles and consisting of well oriented molecu-  
les are usually "bound water". The procedure proposed for  
giving the ratio between the bound water and the quality of  
moist and dry precipitate is based on the assumption that  
electrolytes and non-electrolytes participate in the process  
of molecular adsorption from the aqueous solution of the fine-  
ly dispersed precipitates. This procedure can be used if the  
equilionic sorption of the electrolyte, of which the anion ser-  
ves as indicator, is absent or reduced to a minimum. The  
distribution of the molecules of water and electrolyte ad-  
sorbed by the colloids may be estimated if the electrolyte  
concentration in the intermicellat solution is known.  
There are 3 tables and 11 references, 7 of which are Soviet,  
2 French, 1 American, and 1 English.

Card 2/3

69-58-2 -6/23

The Binding of Water by Finely Dispersed Precipitates 1. Calculation of the Amount of Bound Water From an Electrolyte Concentration in a Solution at Equilibrium.

ASSOCIATION: Institut okeanologii AN SSSR, Moskva (Institute of Oceanology of the USSR Academy of Sciences, Moscow)

SUBMITTED: January 3, 1957

1. Water--Binders 2. Molecules--Adsorption 3. Precipitators  
--Applications 4. Precipitators--Characteristics

Card 3/3

SOV/69-21-4-9/22

5(4)

AUTHOR: Dmitrenko, O.I. and Pavlova, G.A.

TITLE: The Binding of Water by Finely-Dispersed Sediments.  
2. The Dependence of the Amount of Bound Water on the Nature  
and Concentration of Equilibrium Solutions.

PERIODICAL: Kolloidnyy zhurnal, 1959, Vol XXI, Nr 4. pp 419-426 (USSR)

ABSTRACT: This is a study of the water sorption capacity of extramicellar (kaolin) and intermicellar (montmorillonite, beidellite) adsorbents, which are highly-dispersed in equilibrium electrolyte solutions. Sea sediments of the Bering Sea from various depths (95 and 3,400 m) were also investigated. The authors used the anion indicator method. The anions selected for this purpose were  $\text{Cl}^-$ ,  $\text{Br}^-$ ,  $\text{J}^-$  and  $\text{SO}_4^{2-}$ . The assumption of mutual exchangeability of adsorbed electrolyte and water molecules was at the basis of the interpretation of the obtained results [reference 5]. The experiments showed a direct dependence of water dipole adsorption by kaolin on the electrolyte concen-

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SOV/69-21-4-9/22

The Binding of Water by Finely-Dispersed Sediments. 2. The Dependence of the Amount of Bound Water on the Nature and Concentration of Equilibrium Solutions

tration (graph 1). The position of the curves in graph 1 indicates that the quantity of adsorbed water grows in accordance with the following succession of electrolytes:  
 $\text{NaCl} < \text{NaBr} < \text{NaJ} < \text{Na}_2\text{SO}_4$ . This shows, in addition, a direct dependence of the amount of adsorbed water on the solubility of the electrolytes. The experiments with montmorillonite and beidellite, which belong to the intermicellar type of adsorbents (here sorption takes place prevalently in the interior of the disperse particles, and surface adsorption is negligible), also lead to definite results. The amount of absorbed water, however, was found to be inversely proportional to the concentration of the electrolytes (graphs 2 and 3). The authors explain this circumstance with the growing intensity of molecular exchange in increasingly-concentrated solutions of electrolytes, the latter being capable of displacing water dipoles to the extent of the given adsorbent. Also, an inverse proportion of

Card 2/4

SOV/69-21-4-9/22

The Binding of Water by Finely-Dispersed Sediments.2. The Dependence of the Amount of Bound Water on the Nature and Concentration of Equilibrium Solutions.

the amount of absorbed water to the solubility of sodium halide electrolytes could be observed for the two mentioned minerals. Graphs 5 and 6 illustrate the experiments carried out with the above-mentioned sea sediments. In this case potassium halides were used as electrolytes. For both kinds of sediments an inverse proportion of the amount of absorbed water to the electrolyte concentration could again be observed. The curves, however, also show a succession of absorbed electrolytes, which is inverse to the order established for the montmorillonite and beidellite minerals. As the anions are the same and the one difference consists in the size of the ion radiiuses ( $0.98 \text{ \AA}$  for Na and  $1.33 \text{ \AA}$  for K), the obtained results can be explained by enlarged measures of the crystal lattices of adsorbents, which compose the sea sediments. The authors mention the scientist Dzh.D. Bernal

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SOV/69-21-4-9/22

The Binding of Water by Finely-Dispersed Sediments. 2. The Dependence of the Amount of Bound Water on the Nature and Concentration of Equilibrium Solutions.

Reference 17 in their introductory statements.  
There are 6 graphs and 19 references, 9 of which are English,  
5 Soviet, 4 German and 1 French.

ASSOCIATION: Institut okeanologii AN SSSR, Moskva (Institute of Oceanology of the AS USSR, Moscow)

SUBMITTED: 1 February, 1958.

Card 4/4

DMITREJKO, O.I.

Using nonpolarizing cadmium electrodes to study natural electric currents in the sea. Trudy Inst. okean. 35:102-112 '59.  
(MIHA 13:3)

(Electrodes) (Oceanographic instruments)

DMITRENKO, O. I., PAVLOVA, G.A.

Binding of water by highly disperse deposits. Part 3 : Effect  
of the structure of the adsorbents, of the extent of hydration,  
and of the charges of the cations associated with the indicator  
anion. Koll. zhur 22 no.2:154-158 Mr-Ap '60. (MIRA 13:8)

1. Institut okeanologii AN SSSR, Moskva.  
(Adsorption) (Electrolytes)

DMITRENKO, O.I.; RYABININA, A.A.

Reversion of electrolyte adsorption by ferrisilicate and  
aluminosilicate gels. Koll. zhur. 23 no.1:59-66 Ja.-F '61.  
(MIRA 17:2)

1. Institut okeanologii AN SSSR, Moskva.

DMITRENKO, C.I.; PAVLOVA, G.A.

Chemistry of phosphorus in the sea. Part 1. Trudy Inst. okean.  
54:100-114 '62. (MIRA 16:6)  
(Sea water--Analysis) (Phosphorus)

DMITRENKO, O.I.

Bound water in sea sediments in relation to their nature, mechanical composition, quantity of organic carbon, and age variations. Trudy Inst. okean. 54:135-146 '62. (MIRA 16:6)  
(Deep-sea deposits)

DMITRENKO, O.I.

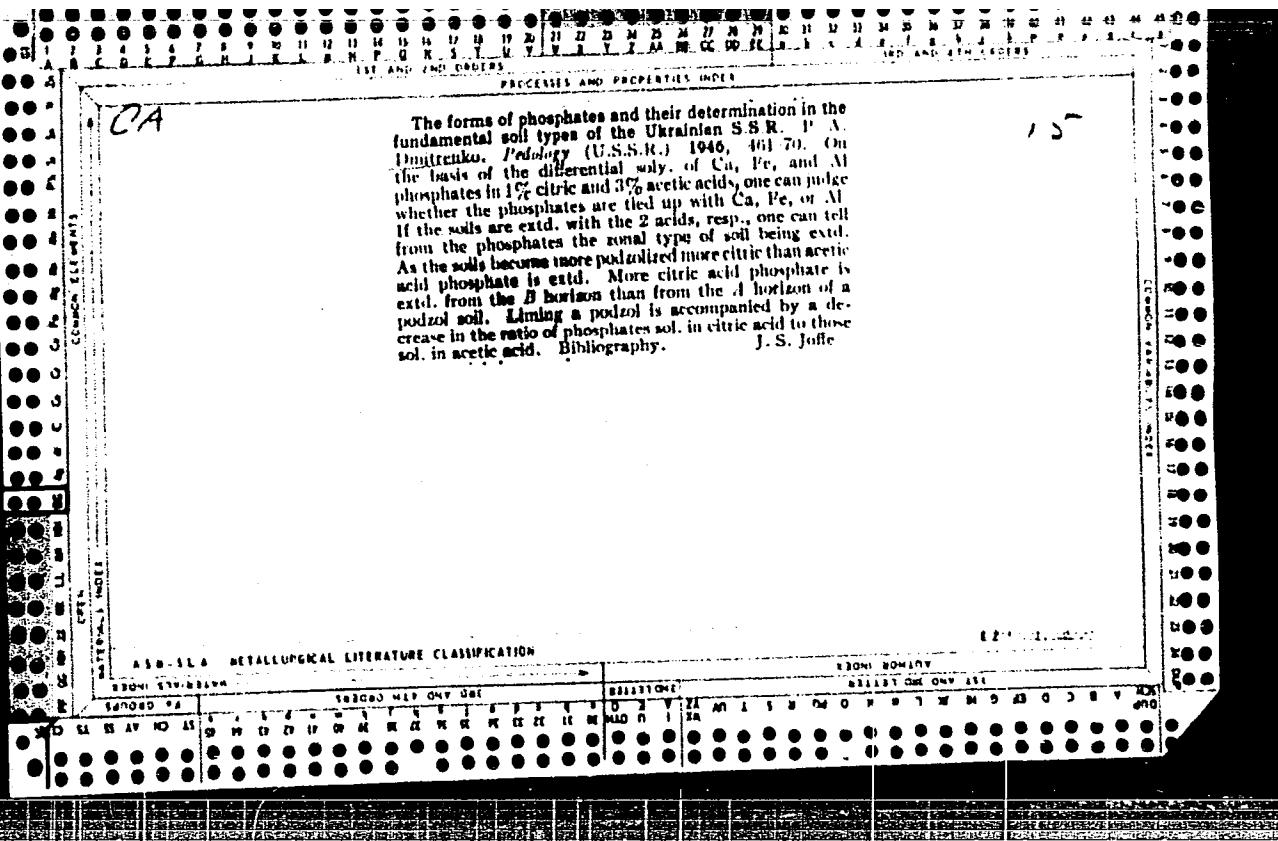
Estimation of changes in the size of molecular capacity (from the adsorption of electrolytes) as an indicator of diagenetic transformations in colloidal sea sediments. Trudy Inst. okean. 54:147-157 '62. (MIRA 16:6)

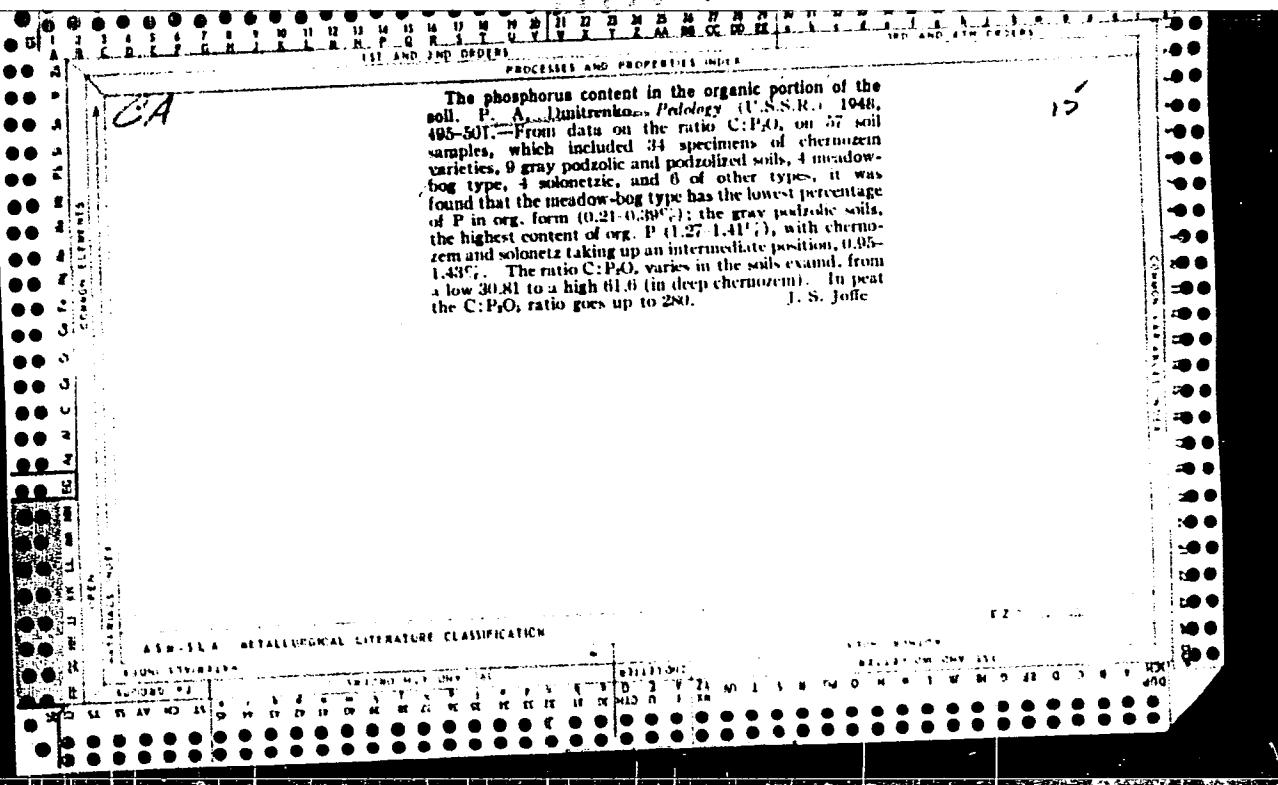
(Deep-sea deposits)

DMITRENKO, P. [Dmytrenko, P.]; ZAKHARCHENKO, P.

State farms deserve high-quality capital construction. Sil'.  
bvd. 12 no.11:6-7 N '62. (MIRA 15:12)

1. Nachal'nik Khar'kovskogo stroitel'no-montazhnogo  
upravleniya tresta "Ukrradgospspetsbud" (for Dmitrenko).  
(Construction industry)





DMITRENKO, P. A.

FA 50/49T14

USSR/Agriculture  
Fertilizers

Oats

May 49

"Significance of Anions in the Feeding of Plants  
With Ammoniacal Nitrogen," P. A. Dmitrenko, Inst.  
Plant Phys and Agr Chem, Acad Sci Ukrainian SSR,  
3½ pp

"Dok Ak Nauk SSSR" Vol LXVI, No 1, pp 85-88.  
Describes experiments on oats in the black earth of  
Novo-Prazhsk Rayon, Kirovogradsk Oblast', by intro-  
ducing two kinds of fertilizer: ammonium chloride  
with anions not absorbed by soil, and ammonium cra-  
late and ammonium citrate with anions absorbed by  
soil. Latter produced positive results. Sub-  
mitted by Acad N. A. Makarov, 4 Mar 49.

USSR/Agriculture (Contd) May 49

50/49T14

*CA**15*

Mobility of phosphoric acid under conditions of slight acidification of the soil. P. A. Dmitrenko. *Doklady Akad. Nauk S.S.R.* 68, 897-900(1949).—The mobility of P in the presence of  $\text{KNO}_3$  of various concn. under the influence of acidification of the soil by various amts. of dil. HCl was examd. P was added in the form of  $\text{KH}_2\text{PO}_4$  (50 mg. per l. per kg. soil) to specimens of typical chernozem soils (from Kirovograd, Kiev, and Zaporozh'e regions). pH variations from 4.1 to 6.3 were examd. After the specimen stood 24 hrs. (10 g. soil and 100 ml. test soln. contg. 0-0.1 N  $\text{KNO}_3$  concn. and enough dil. HCl to give the above range of overall pH values) the clear filtrate was analyzed for phosphate. In all cases increased HCl acidity gave increased mobility of phosphate. However, when increasing amts. of  $\text{KNO}_3$  are added to a given HCl soln. so as to increase soil acidity by this method, the mobility of phosphate declines. Under these conditions the pptn. of Ca phosphates is excluded; hence the effect is caused by adsorption of phosphate ions on colloids of the soil which is enhanced by increased electrolyte ( $\text{KNO}_3$ ) concn. This is proved by decreased mobility of phosphate when NH<sub>4</sub>Cl concn., as the added electrolyte, is gradually increased in concn. G. M. Kosolapoff

IMITRANKO, P.A.; GURTOVENKO, O.O.

Effect of electrolytes on absorption of phosphoric acid by soil. Dopovid Akad. Nauk Ukr. R.S.R. '50, No.5, 387-91 (Russian summary, 391-2).  
(CA 47 no.21:11623 '53) (MLRA 6:4)

DMITRENKO, P. A.

USSR/Biology - Phosphorus  
Fertilizers

11 MAY 50

"Physiological Heterodynamics of Phosphorus for Various Distributions of Fertilizer in the Soil," P. A. Dmitrenko, V. P. Ushakova, Ukrainian Sci Res Inst of Socialist Agr, 3 pp

"Dok Ak Nauk SSSR" Vol LXXII, No 2, pp 397-399.

Discusses results of tests made with millet to find effect on growth of plant, contents of P<sub>2</sub>O<sub>5</sub> in plant, yield of above-ground mass of plant of different distributions of P<sub>2</sub>O<sub>5</sub> in form of Ca(H<sub>2</sub>PO<sub>4</sub>)<sub>2</sub> in soil. Finds it is most effective when placed in upper layer of the soil in layer 4 cm thick -- depth seed is

16077

USSR/Biology - Phosphorus (Contd) 11 May 50

placed. Concludes that not only time but technique of application is important in use of fertilizer. Submitted 20 Feb 50.

1. PODIV, F. A., DMITRENKO, P. A.
2. USSR (600)
4. Chernozem Soils - Ukraine
7. Studing the fertility of the chernozem horizon below plowing depth in the Ukrainian S.S.R., Trudy UNDISOZ 6, 1951.
  
9. Monthly List of Russian Accessions, Library of Congress, May 1953. Unclassified.

1. DMITRENKO, P. A.; GURTOVENKO, O. G.
2. USSR (600)
4. Soil Moisture - Analysis
7. Removing coloring matter from aqueous extracts of solonetz soils. Trudy UNDISOZ 6, 1951.
  
9. Monthly List of Russian Accessions, Library of Congress, May 1953. Unclassified.

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10<sup>-</sup>

Availability of phosphorus for plants in connection with various periods and conditions of interaction of fertilizers with the soil. P. A. Dmitrenko and V. S. Shturanova. *Doklady Akad. Nauk S.S.R.* 76, 447-50 (1951).—A review with 16 references. With meadow-chernozem soil cultures fertilized either directly before planting or 7 months previously, it was shown that with adequate watering no difference in P availability exists between the 2 variants, although the 7-month soil gives somewhat slower initial plant growth. If the soil is 100% satd. with H<sub>2</sub>O, however, the 7-month-old P fertilizer is more available to the plants as shown by more luxuriant growth and P content of the plants. P was used as the 20% P<sub>2</sub>O<sub>5</sub> form of superphosphate. G. M. K.

1957

DMITRENKO, P.A.; SHTURMOVA, V.S.

Peculiarities of the distribution of phosphorus in the horizons of solonetz  
in the south of the Ukrainian S.S.R. *Pochvovedenie* '53, No.2, 60-9.  
(CA 47 no.21:11625 '53) (MLRA 6:3)

1. Ukrain. Sci.-Research Inst. Socialist Agr., Kiev.

Dmitrenko P. O.

S R.

The influence of reduction and oxidation processes upon the conversion of phosphates in the soil. P. G. Dmitrenko and O. V. Boiko. *Dokladi Akad. Nauk Ukrainsk. SSR*, No. 1, 163-7 (Russian Summary). Expts. were made in the lab. with 4 different samples of Ukrainian soil, in which the mobility of the  $\text{P}_2\text{O}_5$  was measured, after it had been added either to a 1% citric acid (1) soln. or in a 3%  $\text{AcOH}$  soln. By measuring the  $\text{Eu}$  in  $\text{MnO}_2$  and the  $\text{FeO}$  and  $\text{MnO}_2$  in the soil, it was established that if the ordinary reducing properties of soil, which normally cause the formation of the lower oxides of Mn and Fe, give way to oxidations, the mobility of the  $\text{P}_2\text{O}_5$  becomes greatly reduced. This phenomenon is much more pronounced in  $\text{AcOH}$  soln. than in 1% soln., because it forms complexes with  $\text{Fe}^{2+}$  and  $\text{Mn}^{2+}$  and  $\text{P}_2\text{O}_5$ , which still show a mobility, whereas  $\text{AcOH}$  leaves the insol.  $\text{FePO}_4$  and  $\text{MnPO}_4$  undissolved in these excesses.

Werner Jacobson

I  
DMYTRENKO, P.O.; SHTURMOVA, V.S.; VLASYUK, P.A., diisnyi chlen Akademiyi nauk URSR.

Effectiveness of phosphorus and nitrogen depending on their location in  
relation to the plant root. Dop. AN URSR no. 4:244-248 '53. (MLBA 6:8)

1. Ukrayins'kyi naukovo-doslidnyi instytut sotszemlerobstva. 2. Akademiya  
nauk URSR (for Vlasyuk).

(Plants, Effect of nitrogen on) (Plants, Effect of phosphorus on)

DMITRENKO, P. A.

Dmitrenko, P. A. -- "Phosphate Condition of Soils in the Ukrainian SSR and Means of Improving Them." Dr Agr Sci, Soil Inst, Acad Sci USSR, 20 Jan 54.. (Vechernyaya Moskva, 8 Jan 54)

SO: Sum 168, 22 July 1954.

18594\* (Effect of Reaction Time of Superphosphate With Soil Upon Availability of Phosphorus to Plants.) Vljanje vremeni reakcijos sitycia supersofata s pochvou na doctupnoj fosforze rastenij. P. A. Dmitrieva and V. S. Shturmova. Fiziko-vedents. 1934, no. 1, p. 3-13.

Tests with cereal crops on different soils. Table, 15 ref.

The utilization of phosphate slags for fertilizer. P. A.  
Dmitriko and T. F. Zavgordnyaya. *Uchbenik i Uraza*  
t. No. 3, 4-4 (1950). In the production of phosphated pig  
iron, the slag of the Martens furnaces carries 13.8-15.8%  
P<sub>2</sub>O<sub>5</sub> and compares very favorably as a source of P with that  
of Thomas slag or superphosphate. In some cases, the slag  
proved superior to a superphosphate. I. S. Jaffe.

Dmitrenko, P.A.

USSR /Chemical Technology. Chemical Products  
and Their Application

I-9

Fertilizers

Abs Jour: Referat Zhur - Khimiya, No 9, 1957, 31298

Author : Dmitrenko P.A., Khudoley I.P.

Title : Suggestion Concerning the Standard of Granulated  
Superphosphate

Orig Pub: Standardizatsiya, 1956, No 4, 68-70

Abstract: On the basis of field tests which showed better  
efficacy of superphosphate of 1-3 mm particle  
size, in comparison with particles of larger di-  
mensions, it is proposed that the existing stand-  
ard be modified and that superphosphate of the  
following granulometric composition be manufac-  
tured: 4-10 mm 3%, 1-4 mm 85%, less than 1 mm 12%.

Card 1/2

USSR /Chemical Technology. Chemical Products  
and Their Application

I-9

Fertilizers

Abs Jour: Referat Zhur - Khimiya, No 9, 1957, 31298

This will relieve the plants of the necessity of  
reprocessing fines into more coarsely granulated  
product and increase the output of the plants.

Card 2/2

DMITRENKO, P. doktor sel'skokhozyaystvennykh nauk.; SKORINA, S.A.

Organize the study of soil on collective and state farms. Zemledelie  
4 no.10:116 0 '56. (MIRA 9:11)  
(Soils)

DMITRENKO, P.A.; SHTURMOVA, V.S.

Effect of the size of superphosphate granules on the availability  
of phosphorus to plants. *Pochvovedenie* no.10:33-35 O '56.

(MDRA 10:1)

1. Ukrainskiy nauchno-issledovatel'skiy institut sotsialisticheskogo  
zemledeliya.

(Phosphates) (Plants--Nutrition)

J

Country : USSR  
Category: Soil Science. Mineral Fertilizers.

Abs Jour: RZhBiol., No 18, 1958, № 82116

Author : Dmitrenko, P.A.  
Inst : -  
Title : Availability of Phosphorus to Plants in Respect to  
Different Conditions of Interaction of Fertilizer with  
Soil.

Orig Pub: Pochvovedeniye, 1957, № 6, 11-15

Abstract: The Ukrainian Scientific Experimental Institute conducted field and vegetative experiments applying the method of marked atoms of P. According to the results of the experiments (figures not presented), phosphates introduced into chernozem and similar soils were found for 6-7 months in an assimilable

Card : 1/2

- - -

Country : USSR  
Category: Soil Science. Mineral Fertilizers.

J

Abs Jour: RZhBiol., № 18, 1958, No 82116

form, and absence of their after-effect was explained not by the retrogradation of  $P_2O_5$  but by an insufficiency of N and K used by the 1st culture. Numerical data are presented which show the advantage of tiny granules of  $P_c$  over large ones. The agricultural cultures are enumerated which reacted negatively to the addition of lime to P. -- N.N. Sokolov

Card : 2/2

USSR / Soil Science. Mineral Fertilizers.

J-4

Abs Jour: Ref Zhur-Biol., No 8, 1958, 34386.

Author : Dmitrenko, P. A., Kondratenko, E. S.

Inst : Not given.

Title : About Zoning and Utilization of Fortilizers Along  
the Ukraine.

Orig Pub: Udobreniye i urozhay, 1957, No 7, 5-13.

Abstract: No abstract.

Card 1/1

USSR/Soil Science - Mineral Fertilizers.

J.

Abs J<sup>O</sup>ur : Ref Zhur - Biol., № 15, 1958, 67957

Author : Dmitrenko, P.D.

Inst : Ukrainian Scientific Research Institute of Agriculture.

Title : Granulated Superphosphate (A Discussion).

Orig Pub : Udobreniye i urozhay, 1957, № 12, 22-29.

Abstract : Experiments at the Ukrainian Scientific Research Institute of Agriculture have demonstrated that it is better to use  $P_c$  in small (0.5-3 mm.) than in large (5-7 mm.) granules. When large-granule  $P_c$  was used, less P entered the plant in the initial growth period; in addition, the plants grew at different rates since those near the granules got excessive amounts of phosphorous. This occurred not only when  $P_c$  in large granules was added to the rows but also when it was scattered broadside. When small-grained  $P_c$

Card 1/2

USSR/Soil Science - Mineral Fertilizers.

J.

Abs JOur : Ref Zhur - Biol., No 15, 1958, 67957

was used, there was no such differentiation in plant growth; all plants were equally supplied with phosphorous, especially important in the initial period of growth. If the quantity of granules less than 2 mm. in size in the  $P_c$  is increased, it will simplify the process of producing this fertilizer, raising the productivity of superphosphate factories by 25%. To apply such  $P_c$  it is necessary to use combine sowers which plant the seed and apply the fertilizer separately, thus eliminating the possibility that the acidity of the  $P_c$  will have an adverse effect on germination. -- O.P. Medvedeva

Card 2/2

- 45 -

USSR/Soil Science. Physical and Chemical Properties of J  
Soils

Abs Jour : Ref Zhur-Biol., No 13, 1958, 58286, By C.A.Nikitin

Author : Dmitrenko P. A.  
Inst : Soil Institute, Academy of Sciences USSR  
Title : Phosphate Regime in the Soils of Ukrainian SSR  
and Methods for its Improvement (1529-374)

Orig Pub : Tr. Pochv. in-ta, AN SSSR, 1957, 50, 122-274

Abstract : The forms of phosphorus compounds and the characteristics of the distribution of P according to the horizons in the main types of soil on the territory of Ukrainian SSR, the main laws of the seasonal dynamics of P in the soils under conditions of the effective application of phosphorus fertilizers are discussed. Examined are in detail the characteristics of the phosphorus

Card 1/4

USSR/Soil Science. Physical and Chemical Properties of J  
Soils

Abs Jour : Ref Zhur-Biol., No 13, 1958, 58286, By C.A.Nikitin

Abstract : regime in the more widespread and important from a production point of view soils: chernozems, gray forest-podzol soils, peat-podzol soils, saline soils of Polcs'ye, forest-steppe, soils, and soils in the Southern Rayons of Ukrainian SSR. Chernozems contain an average of 0.5% of P; dark-chestnut soils--0.3 to 0.5%; saline soils--0.2 to 0.4%. Plowed soil horizons being biologically more active produce the largest quantities of mobile P under aerobic decomposition of the organic substances. As compared with chernozems, the podzolized and podzol soil with an acid reaction contain smaller quantities of phosphate of calcium which are soluble in acetic acid. In gley and illuvial horizontal saline and podzol soils the role of phosphates with

Card 2/4

USSR/Soil Science. Physical and Chemical Properties of J  
Soils

Abs Jour : Ref Zhur-Biol., No 13, 1958, 58286, By C.A.Nikitin

Abstract : sesquioxides increases. The activation of biological processes in soils contributes to the involvement of a large number of primary forms of mineral slightly soluble phosphates into the rotation circuit. In the lower horizons of saline soils P, notwithstanding its greater solubility, cannot be used by the plants because of the large accumulation of salts in the soil. An acid reaction and an oxidation-reduction processes in the soils contribute to the decrease in P motility. The poor effectiveness of phosphorus fertilizers in some soils is conditioned by the formation of difficultly soluble P compounds in the soils, and by the insufficiency of N and K in these soils. A gradual decrease in the solubility of phosphates was observed

Card 3/4

USSR/Soil Science. Physical and Chemical Properties of J  
Soils

Abs Jour : Ref Zhur-Biol., No 13, 1958, 58236, By C.A.Nikitin

Abstract : when the period of interaction of fertilizers  
and peat-podzol soils was prolonged to 1½  
years. Conditions of the formation of "t.n."  
? physiologically incompletely usable P are  
examined. The use of phosphorite fertilizers  
in combination with nitrogen-potassium ferti-  
lizer and granulated phosphorus fertilizers  
is recommended.

Card 4/4

VLASYUK, Petr Antipovich, akademik; KOSMATYY, Yevdokim Stepanovich,  
kand.khim.nauk; DMITRENKO, P.A., otv.red.; SKOL'ZNEVA, Ye.A.,  
red.; MANOYLO, Z.T., khudosn.-tekhn.red.

[Tagged atom method in agricultural physiology] Metod mechenykh  
atomov v agrofisiologii. Kiev, Izd-vo Ukrainskoi akad.sel'khoz.  
nauk, 1959. 326 p. (MIRA 13:5)

1. Vsesoyuznaya akademiya sel'skokhozyaystvennykh nauk imeni V.I.  
Lenina; AN USSR; Ukrainskaya Akademiya sel'skokhozyaystvennykh  
nauk (for Vlasuk). 2. Chlen-korrespondent Ukrainskoy akademii  
sel'skokhozyaystvennykh nauk (for Dmitrenko).  
(Tracers (Biology)) (Plant physiology--Research)

DMITRENKO, P.A.; SHEVCHENKO, L.A.

Intensifying the aftereffect of phosphorus fertilizers.  
Pochvovedenie no.7:74-80 '60. (MIRA 13:7)

1. Ukrainskiy nauchno-issledovatel'skiy institut zemledeliya.  
(Fertilizers and manures)  
(Plants, Effect of phosphorus on)

SHVETS, Ivan Trokhimovich [Shvets', I.T.]; OVCHARENKO, Feodor Danilovich, akademik; DOBROKHOTOV, Nikolay Nikolayevich [Dobrokhotov, M.M.], akademik, zasluzhennyy deyatel' nauki i tekhniki USSR; STUDENNIKOV, Timofey Vasil'yevich [Studennykov, T.V.]; BAKUMA, Pavel Fedorovich, akademik; DMITRENKO, Petr Alekseyevich [Dmytranko, Petro Oleksiovych]

Congress of conquerors. Znan. ta pratsia no.10:1-5 O '61.  
(MIRA 14:8)

1. Rektor Kiyevskogo gosudarstvennogo universiteta im. T.G. Shevchenko (for Shvets). 2. AN USSR (for Ovcharenko).
3. Nachal'nik upravleniya transporta i svyazi Ukrainskogo sovnarkhoza (for Studennikov). 4. Chlen-korrespondent Ukrainskoy Akademii sel'skokhozyaystvennykh nauk (for Dmitrenko).  
(Russia—Economic conditions)

DMITRENKO, P.A. [Dmytrenko, P.O.]; GOLOVASHCHUK, Zh.T. [Holovashchuk,  
Zh.T.]

Effect of various forms of mineral fertilizers on the starch and  
ascorbic acid content of potato tubers. Ukr.biokhim.zhur. 34  
no.6:863-870 '62. (MIRA 16:4)

1. Ukrainian Research Institute of Agriculture, Kiev.  
(POTATOES--FERTILIZERS AND MANURES) (STARCH) (ASCORBIC ACID)

DMITRENKO, P.A.; SHESTIDESYATNAYA, N.Ye.

Effect of liming on soil fertility in Transcarpathia. *Pochvovedenie*  
no.10:40-46 O '62. (MIRA 15:11)

1. Ukrainskiy nauchno-issledovatel'skiy institut zemledeliya.  
(Transcarpathia--Liming of soils)  
(Transcarpathia--Soil fertility)

DMITRENKO, P.A.; DIDYCHENKO, A.P.

Effectiveness of single and multiple placement of mineral fertilizers in soils. Pochvovedenie no.8:73-82 Ag '63.  
(MIRA 16:9)  
1. Ukrainskiy nauchno-issledovatel'skiy institut zemledeliya.

DMITRENKO, P.A.; TOMASHEVSKAYA, Ye.G.; SHTURMOVA, V.S.

Characteristics of the phosphorus nutrition of cereal and  
leguminous plants at the beginning of their growth. Fiziol.  
rast. 10 no.2:142-147 Mr-Ap '63. (MIRA 16:5)

1. Ukrainian Scientific Research Institutes of Agriculture.  
(Plants, Effect of phosphorous on) (Grain) (Legumes)

DMITRENKO, Petr Alekseyevich[Dmytrenko, P.O.]; SHCHEDRIN, V.V.,  
red.

[Keys to fertility; ways for the effective use of mineral  
fertilizers in the Ukraine] Kliuchi do rodiuchosti; shliaky  
efektyvnoho zastosuvannia mineral'nykh dobryv na Ukraini.  
Kyiv, 1965. 45 p. (Tovarystvo "Znannia" Ukrains'koi RSR.  
Serija VIII, no.9) (MIRA 18:9)

1. Chlen-korrespondent AN Ukr.SSR (for Dmitrenko).

DMITRENKO, P.A.; VITRIKHOVSKIY, P.I.

Different ability of various legumes to assimilate phosphorus  
from hard-to-dissolve phosphates. Dokl. Akad. sel'khoz. nauk  
no. 3:22-23 Mr '65. (MIRA 18:5)

1. Ukrainskiy nauchno-issledovatel'skiy institut zemledeliya.
2. Chlen-korrespondent AN UkrSSR (for Dmitrenko).

DMITRENKO, P.A. [Dmytrenko, P.O.]; LUGOVSKAYA, Ye.Ya. [Luhovs'ka, K.IA.];  
TOMASHEVSKAYA, Ye.G. [Tomashevs'ka, O.H.]

Characteristics of the nutrition of grain crops and legumes in  
their mixed sowing. Dop. AN URSR no.9:1225-1228 '65.

(MIRA 18:9)

1. Ukrainskiy nauchno-issledovatel'skiy institut zemledeliya.
2. Chlen-korrespondent AN UkrSSR (for Dmitrenko).

DMITRENKO, P.F. (st. Poltava-Kiyevskaya)

Repair and maintenance of road beds. Put.i put.khoz. no.4:32-33  
Ap '57. (MLRA 10:5)

1.Nachal'nik Poltava-Kiyevskoy distantsii.  
(Railroads--Maintenance and repair)

1. Ukraine, Kyiv naukovo-praktichnyi institut sotsialisticheskogo zemle-

zobetva. Predstavlyayushchiy ATUSSR P.A. Vlasjuk.

(Solonets'kyy soli) (Gypsum) (Phosphate)

containing sodium carbonates. Dop. ATUSSR no. 6:594-597, 56.

Effect of gypsum on the formation of phase phases in solonets'kyy

(MIRA 10:2)

Dmitriko, P.O.: SPIVAK, M.S.

1. Ukrains'kyj naukovo-doslidnyj instytut zemelerobstva.

1964. 137 p.  
udobrennia si, skohospodars'kykh kul'tur. Kieve, Urozhai,  
[Nutrition and fertilizers of farm crops] Zhyvlenija ta

nak. red.; YASHOVSKY, I.V. [IAshovs'kyi, I.V.], red.  
S. A. [Samotsavch, S.A.], red.; Pmyrova, N.A., kand. sel'khoz.  
F. A., red.; SAMBUR, G.M. [Sambur, H.M.], red.; SAMSEVICH,  
[Lazurs'kyi, O.V.], red.; KUKSI, M.V., red.; LAZURSKY, O.V.  
FOLDOV, O.M., red.; IL'yASHENKO, M.G. [IL'iashenko, M.H.], red.;  
nak, red.; II.YASHENKO, M.G. [IL'iashenko, M.H.], red.;  
red.; DMITRENKO, P.O. [Dmytrenko, P.O.], doktor sel'khoz.  
[Vysyns'kyi, O.M.], red.; GOLOMBA, R.A. [Holomba, R.A.]  
YUKHINCUK, F.P. [Iukhmejuk, F.P.], ott. red.; VISHINSKY, O.M.

1. Ktev. Ukrayins'kyi naukovo-doslidnyi instytut zemelerodstva.  
nitsa. Kyiv, Urozhai, 1964. 160 p. (MIRA 17:12)  
[Fertilizers and cultivation practices] Dobryva ta agrotekhnika.  
KATRENNKO, K.A., red.  
SANTSEVICH, S.A. [Samstsevych, S.A.], red.; FEODOROVA, N.A., red.;  
POPOV, F.A., red.; SAMBUR, G.M. [Sambar, H.M.], red.;  
KUKSIN, M.V., red.; LAZURSKIY, O.V. [Lazurs'kyi, O.V.], red.;  
N.G. [IL'IAshenko, N.H.], red.; KOLOBOV, O.M., red.;  
red.; DMITRENNKO, P.O. [Dmytrenko, P.O.], red.; IL'IAshenko,  
[Vyshtyns'kyi, O.M.], red.; GOLOMBA, R.A. [Holomba, R.A.],  
YUKHIMCHUK, F.P. [Ukhymchuk, F.P.], otyv. red.; VISHINSKIY, O.M.

1. Glavnyy Inzh. proyektta Gosudarstvennogo proyektno-konstruktorskogo i nauchno-issledovatel'skogo instituta morskogo transporta MInstterstva morskogo flota SSSR (for Medovikov). 2. (osudarstvennyy konstruktorskyy i nauchno-issledovatelskiy institut morskogo transporta MInstterstva portov MInstterstva morskogo flota SSSR (for Dmitrenko).

(MIRA 18:5)

A new port in the Gulf of Aden. Mor. flot 25 no. 5:43 My 165.

MEDOVIKOV, I.; DMITRENKO, S., Inzh.

techn 86 no. LC 56 18 Ap 165

If the profession of a graduate engineer a creative one? Przegej

Wladyslaw, mgr inz.; BRONIKOWSKI, Adam; STASIKOWSKI, Henryk  
Gican, Ryszard, mgr inz.; Dlugoszko, Stefan, mgr inz.; LASKOWSKI,  
Kosciejkiewicz, Adam, mgr inz.; MALINOWSKI, Kazimierz, mgr inz.;  
CHYCIK, Andzej, mgr inz.; LASKOWSKI, Wladyslaw; SOWA, Zbigniew, mgr inz.

(Agricultral Laborers)

(MIRA 15:5)

1962. 278 p.

Sbornik dokumentov. Moskva, Izd-vo TSK VIKSM "Molodets na seliine".  
Vtorgin lads] V krai prository i podvijovi; Molodets na seliine.

[In the land of wide-open spaces and heroic deeds; youth in the

red.

FEDOROV, A.G., red.; IZUBUSHKINA, Ye., red.; YEGOROVA, I., tekhn.  
RAKHMANIN, B.P.; BORISOV, Yu.S., oty. red.; KRUCHINA, N.Ye., red.;  
TEL'PUKHOVSKIY, V.B.; DMITRENKO, T.A.; ZELENIN, I.Ye.; KOSTYAKOVA, G.K.;

DMITRENKO, P.F. (st. Poltava-Kiyevskaya)

Repair and maintenance of road beds. Put.i put.khoz. no.4:32-33  
Ap '57. (MLRA 10:5)

1.Nachal'nik Poltava-Kiyevskoy distantsii.  
(Railroads--Maintenance and repair)

DMITRENKO, P.F. (st. Poltava-Kiyevskaya)

Repair and maintenance of road beds. Put.i put.khoz. no.4:32-33  
Ap '57. (MLRA 10:5)

1.Nachal'nik Poltava-Kiyevskoy distantsii.  
(Railroads--Maintenance and repair)

DMITRENKO, P.O.; SPIVAK, Kh.S.

Effect of gypsum on the formation of phosphates in solonetz soils containing sodium carbonates. Dop. AN URSR no. 6:594-597 '56.  
(MLRA 10:2)

1. Ukrains'kiy naukovo-doslidnyi institut sotsialistichnogo zemle-robstva. Predstaviv akademik AN URSR P.A. Valsyuk.  
(Solonetz soils) (Gypsum) (Phosphates)

YUKHIMCHUK, F.P.[IUkhymchuk,F.P.], otv. red.; VISHINSKIY, O.M.  
[Vyskyns'kyi, O.M.], red.; GOLOMBA, R.A.[Holomba, R.A.]  
red.; DMITRENKO, P.O.[Dmytrenko, P.O.], doktor sel'khoz.  
nauk, red.; IL'YASHENKO, M.G.[Illiashenko, M.H.], red.;  
KOLOBOV, O.M., red.; KUKSIN, M.V., red.; LAZURSKIY, O.V.  
[Lazurs'kyi, O.V.], kand. sel'khoz. nauk, red.; POPOV,  
F.A., red.; SAMBUR, G.M.[Sambur, H.M.], red.; SAMTSEVICH,  
S. A. [Samtsevych, S.A.], red.; FEDOROVA, N.A.,kand.sel'khoz.  
nauk. red.; YASHOVSKIY, I.V.[IAshovs'kyi, I.V.], red.

[Nutrition and fertilizers of farm crops] Zhyvlennia ta  
udobrennia sil's'kohospodars'kykh kul'tur. Kiev, Urozhai,  
1964. 137 p. (MIRA 17:10)

1. Ukrains'kyj naukovo-doslidnyy instytut zemlerobstva.

YUKHIMCHUK, F.P.[IUkhymchuk, F.P.], otv. red.; VISHINSKIY, O.M.  
[Vyshyns'kyi, O.M.], red.; GOLOMBA, R.A.[Holomba, R.A.],  
red.; DMITRENKO, P.O.[Dmytrenko, P.O.], red.; IL'YASHENKO,  
M.G.[Illiashenko, M.H.], red.; KOLOBOV, O.M., red.;  
KUKSIN, M.V., red.; LAZURSKIY, O.V.[Lazurs'kyi, O.V.], red.;  
POPOV, F.A., red.; SAMBUR, G.M.[Sambur, H.M.], red.;  
SAMTSEVICH, S.A.[Samtsevych, S.A.], red.; FEDOROVA, N.A., red.;  
KATRENKO, K.A., red.

[Fertilizers and cultivation practices] Dobryva ta agrotekhnika. Kyiv, Urozhai, 1964. 160 p. (MIRA 17:12)

1. Kiev. Ukrains'kyi naukovo-doslidnyi instytut zemlerobstva.

MEDOVIKOV, I. & DMITRENKO, S., inzh.

A new port in the Gulf of Aden. Mor. flot 25 no. 5:43 My '65.  
(MIRA 18:5)

1. Glavnnyy inzh. proyekta Gosudarstvennogo proyektno-konstruktorskogo i nauchno-issledovatel'skogo instituta morskogo transporta Ministerstva morskogo flota SSSR (for Medovikov). 2. Gosudarstvennyy proyektno-konstruktorskiy i nauchno-issledovatel'skiy institut morskogo transporta Ministerstva morskogo flota SSSR (for Dmitrenko).

CHYCKI, Andrzej, inż.; LASKOWSKI, Włodysław,; SOWA, Zbigniew, mgr inż.;  
KOSCIELNIAK, Adam, mgr inż.; MALINOWSKI, Kazimierz, mgr inż.;  
CYGAN, Ryszard, mgr inż.; DMITRENKO, Stefan, mgr inż.; LASKOWSKI,  
Włodysław, mgr inż.; BRONIKOWSKI, Adam; STASIKOWSKI, Henryk

Is the profession of a graduate engineer a creative one? Przegl  
techn 86 no. 185 18 Ao '65

TEL'PUKHOVSKIY, V.B.; DMITRENKO, T.A.; ZELENIN, I.Ye.; KOSTYAKOVA, G.K.; RAKHNMAMIN, B.P.; BORISOV, Yu.S., otv. red.; KRUCHINA, N.Ye., red.; FEDOROV, A.G., red.; LYUBUSHKINA, Ye., red.; YEGOROVA, I., tekhn. red.

[In the land of wide-open spaces and heroic deeds; youth in the virgin lands] V kraiu prostorov i podvigov; molodezh na tseline. Sbornik dokumentov. Moskva, Izd-vo TsK VLKSM "Molodaia gvardiia," 1962. 278 p.

(MIRA 15:5)

(Agricultural laborers)

SHUL'KEVICH, Mikhail Mironovich,; DMITRENKO, Talyda Danilovna,; ZAYCHENKO,  
R., red.; GARSHANOV, A.,red.; IOAKIMIS, A.,tekhn. red.

[Kiev; with an index to the map]. Kiev; ukazatel' k plan-skhemе.  
Kiev, Gos. izd-vo lit-ry po stroit.i arkhit. USSR, 1958. 45 p.  
1 fold map. (MIRA 11:12)

(Kiev--Maps)

"APPROVED FOR RELEASE: 06/12/2000

CIA-RDP86-00513R000410510014-5

Dmitrukko, T.D.

SHUL'KEVICH, Mikhail Mironovich; DMITRUNKO, Talida Danilovna; ZAYCHENKO, R.,  
red.; GARSHANOV, A., red.; IOAKIMIS, A., tekhn. red.

[Kiev; guide to the map] Kyiv; pokazhchik do plam-skhemy. Kyiv,  
Derzh. vyd-vo lit-ry z budivnytstva i arkhit. UkrSSR, 1958. 86 p.  
(Kiev--Maps) (MIRA 11:7)

APPROVED FOR RELEASE: 06/12/2000

CIA-RDP86-00513R000410510014-5"

DMITRENKO, T. K., Cand Med Sci -- (diss) "Problem of the mechanism of the action of some anti-thyroid substances." Alma-Ata, 1960. 16 pp; (Joint Academic Council of the Inst of Physiology, Inst of Kray Pathology, and the Inst of Clinical and Experimental Surgery of the Academy of Sciences Kazakh SSR); 200 copies; price not given; (KL, 28-60, 164)

KROPACHEV, V.; DMITRENKO, V., starshiy inzh.

Modernization of the gantry crane "Gants." Rech. transp. 21  
no.3:13-15 Mr '62. (MIRA 15:4)

1. Nachal'nik otdela mekhanizatsii portaimeni Lenina, Dneprovskogo  
basseyna (for Kropachev). 2. Otdel mekhanizatsii porta imeni  
Lenina Dneprovskogo basseyna (for Dmitrenko).  
(Cranes, derricks, etc.)

DMITRENKO, V. A.

188T4

USSR/Chemistry - Geology

Jul/Aug 51

"Artificial Weathering and Synthesis of Minerals Under Electrodialysis," O. I. Dmitrenko, V. A. Kargin, Inst of Geol Sci, Acad Sci USSR, Dept of Sedimentary Petrography

"Kolloid Zhur" Vol XIII, No 4, pp 259-266

First publication of original method of low-temperature electrosynthesis of minerals of the weathered crust allowing to conduct expts on synthesis of minerals at concns much less than 1 mg/l of water under normal temp and pressure without addn of any chem reagents. Five-chamber electrodialysis app permits reproduction of natural weathering of aluminum silicate.

188T4  
USSR/Chemistry - Geology (Contd)

Jul/Aug 51

cates, in which the usual hydrolysis is speeded up many times. By varying the conditions, cryst varieties never or seldom found in nature can be produced, especially in the case of compds with large energy of crystn. Method can be used in biology, medicine, crystallography, crystal chemistry, soil science, and lithology.

188T4

DMITRENKO, V. I.:

DMITRENKO, V. I.: "The dynamics of chemical factors of nervous excitation in patients with cancer of the cervix uteri." Min Health Ukrainian SSR. Khar'kov Medical Inst. Khar'kov, 1956  
DISSERTATION For the Degree of Candidate in Medical Science.

So: Knizhnaya 'etopis', No. 18, 1956

Dmitriyenko, V.I.

USSR/General Problems of Pathology - Tumors.

T-5

Abs Jour : Ref Zhur - Biol., No 3, 1958, 12692

Author : Dmitriyenko, V.I.

Inst : Not given

Title : Chemical Mediators in the Nervous System in Patients  
with Carcinoma of the Cervix.

Orig Pub : V sb.: Vopr. luchevoy terapii. Kiyev, Gosmedizdat USSR,  
1956, 113-122

Abstract : Acetylcholine content and cholinesterase activity were  
studied in 92 patients with carcinoma of the cervix,  
before and after radiation therapy. In 2/3 of the pa-  
tients, irrespective of the stage of the disease, there  
were deviations from the normal as far as the contents  
of acetylcholine and cholinesterase were concerned;  
this deviation consisted of an increase in both subs-  
stances

Card 1/2

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YAKOVLEV, Ye.I., dotsent (Khabarovsk)

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IMITRENKO, V.I., dtsent

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